Evaluations of Antioxidant Effects of Selected Medicinal Plant Extracts Claimed to Treat Kidney Stone Disease

Tilahun Alelign^{1,*}, Asfaw Debella², Beyene Petros¹

ABSTRACT

Introduction: Free radicals are capable of inducing oxidative damage, which causes various human diseases. However, antioxidants reduce the risk of diseases related to reactive oxygen species. Medicinal plants such as Achyranthes aspera leaves, Satureja punctata aerial parts, Aloe pulcherrima gel, Gomphocarpus fruticosus leaves and Commiphora myrrha resins were claimed to treat various ailments including urolithiasis in Ethiopia. Objectives: This study was intended to determine phytochemicals and antioxidant activities of the aforementioned plants. Materials and Methods: Plants were collected and the aqueous crude extracts were prepared. Phytochemicals were screened qualitatively and DPPH(2,2-diphenyl-1-picrylhydrazyl) radical scavenging assay was measured at 517nm using UV-Vis spectrophotometry. Data were analyzed statistically using one-way ANOVA, Dunnett's comparison test of the Graph Pad Prism version 6. Results: The plant extracts exhibited various phytochemicals such as phenols, flavonoids and tannins, while these were absent in C. myrrha. Steroids and terpenoides were absent in A. pulcherrima and G. fruticosus extracts, respectively. DPPH scavenging capacities of S. punctata, G. fruticosus, A. pulcherrima and A. aspera aqueous extracts were 92.3%, 81.6%, 72.3% and 54.9%, respectively compared to control (Ascorbic acid) showed 87.6%, 94.5%, 92.3% and 95.6%, respectively at inhibitory concentrations of 0.20 mg/ml, 0.78 mg/ml, 3.13 mg/ml and 12.5 mg/ml, respectively. The IC_{50} antioxidant values of *S. punctata*, *A. pulcherrima*, *G. fruticosus*, and *A. aspera* extracts were 0.01 ±0.003 mg/ml, 0.42 ±0.047 mg/ml, 1.64 ±0.147 mg/ml, and 13.51 ±1.08 mg/ml, respectively compared to Ascorbic acid (0.03 ±0.007 mg/ml). Conclusion: The phytoconstituents in S. punctata aqueous extract has the best capability to scavenge DPPH free radicals. Future characterizations of compounds responsible for the antioxidant activities will be required.

Key words: Antioxidant activity, DPPH, Free radicals, Medicinal plants, Phytochemicals.

INTRODUCTION

Excess oxygen could result in oxidative damage, which may lead to death due to its role in the reduction of certain products into free radicals. The formation of free radicals (highly reactive oxygen species) through the activity of the respiratory chain in the mitochondria is part of the cell's normal metabolic process, including detoxification process and immune defenses.1 Oxidative stress can be initiated if the generation of free radicals exceeds the body's ability to protect the effect of antioxidants. The excessive generation of free radicals such as nitric oxide and peroxynitrite (an unstable structural isomer of nitrate),² superoxide anions, hydrogen peroxides and hydroxyl radicals are capable of inducing oxidative damage to the body.^{1,3} Radicals cause damage to various cellular macromolecules such as DNA, proteins and lipids, which trigger various human chronic diseases.^{2,4} However, the proper physiological functions of the body could be maintained by bodies endogenous antioxidant systems,⁵ and by ingestion of exogenous antioxidants.⁴ Therefore, adequate antioxidant defense within and outside cells is important to offer protection against oxidative damage.⁶

In aerobic organisms, the defense system against free radicals is provided by scavengers, which act as antioxidants. This functions by donating an electron to free radicals, thereby, the paired electrons stabilized. Human cells have an array of protecting mechanisms by preventing the production of free radicals through enzymatic and non-enzymatic antioxidants. The first antioxidant defense may involve enzymes such as superoxide dismutase (SOD), catalase (CAT) and glutathione peroxidase (GPX); whereas the non-enzymatic mechanisms include nutrients and minerals.7,8 The glutathione (GSH) detoxifies hydroxyl radicals.9 The activities of antioxidant enzymes were reduced due to defects in the antioxidant defense system.10 However, the antioxidant constituents of plants might restore

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Debella², Beyene Petros¹ ¹Addis Ababa University, Addis Ababa, FTHIOPIA

Tilahun Alelign^{1,*}, Asfaw

²Ethiopian Public Health Institute, Traditional and Modern Drug Research, ETHIOPIA.

Correspondence

Dr. Tilahun Alelign

Addis Ababa University, Addis Ababa, ETHIOPIA.

Phone no: +251-912038343

E-mail: tilaalelign@gmail.com

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the renal antioxidant enzymes, which protect from renal cell injury.¹¹ The non-enzymatic antioxidants can be classified as water-soluble and lipid-soluble depending on whether they act primarily in the aqueous phase or in the lipophilic region of the cell membrane. Ascorbic acid (vitamin C) is a naturally occurring antioxidant found in medicinal plants, vegetables, fruits and whole grains. The hydrophilic antioxidants include Vitamin C and certain polyphenol flavonoid groups; whereas the lipophilic antioxidants include ubiquinols, retinoids, carotenoids, apocynin and procyanidins.¹² DPPH (2,2-diphenyl-1-picrylhydrazyl) is an organic compound, which is a dark-red colored crystalline powder composed of free-radical molecules.¹³ DPPH assay is a widely used model to evaluate antioxidants because of its ease of use and reproducibility.14,15 In the DPPH assay, the initial electron transfer occurs very quickly and the subsequent hydrogen transfer occurs more slowly and depends on the hydrogen-bond accepting solvents. DPPH free radical scavenging effects of antioxidants is due to their hydrogen atom-donating ability.^{16,17} In antioxidant assays, a strong hydrogen-bond accepting solvents such as methanol and ethanol are usually used.¹⁸ Electron transfer assays measure the reducing ability of the substrate (antioxidant), whereas the hydrogen transfer assays measure hydrogen donating ability of the substrate. Previous studies reported that assays that measure hydrogen transfer would be preferable to assays that measure electron transfer.¹⁹ If free radicals have been scavenged by plant extracts, DPPH solutions could be changed from violet to yellow color; which is a decolorization effect.20

Absorbance can be measured after the reactions of antioxidants with DPPH, which are dissolved in methanol or ethanol.¹³ DPPH radical has a maximum UV-Vis spectrophotometry absorption in the range between 515nm and 519nm.^{13,21} The reduction in absorbance of DPPH radicals in the solution revealed the actions of antioxidant agents.^{17,22} Therefore, the ability of the plant extracts to scavenge DPPH free radicals could contribute to the inhibition of the inflammatory processes.²³ The polyphenol of plant extracts is an excellent antioxidant, which act by hydrogen donation of their phenolic hydroxyl group to stop free radical chain reactions.²⁴ The synthetic antioxidants such as butylated hydroxytoluene and butylated hydroxyanisole have recently been reported to be dangerous for human health.⁴ Therefore, searching an effective, non-toxic and natural antioxidant could be an effective treatment option to manage renal oxidative stress.

In Ethiopia, medicinal plants like *Achyranthes aspera* L. and *Satureja punctata* (Benth.) Briq.^{25,26} *Aloe pulcherrima*-Gilbert and Sebsebe;²⁶ *Gomphocarpus fruticosus* (L.) Ait.f, (key informant); and *Commiphora myrrha* (Nees) Engl.²⁶ have been claimed to treat various diseases including kidney stones. Before the commencement of this work, the insufficiency of scientific literature was assessed in relation to phytochemical analysis and antioxidant effects. Therefore, the present study was aimed at providing information on phytochemicals and the *in vitro* DPPH scavenging activities of the aforementioned plant extracts.

MATERIALS AND METHODS

Medicinal Plant Collection and preparation of Crude Extracts

The A. aspera L. leaves were collected from Addis Ababa University "Arat kilo" Campus; S. punctata (Benth.) Briq. aerial parts at "Entoto" forest region, Northwest to Addis Ababa; and A. pulcherrima Gilbert and Sebsebe gel were collected from Ankober District of North Shewa Zone. the leaf of Gomphocarpus fruticosus (L.) Ait.f (Asclepiadaceae) collected at Bole Bulbula around "93 Mazoria" Addis Ababa; whereas the resins of Commiphora myrrha (Nees) Engl. (Burseraceae) were purchased from Merkato market, Addis Ababa. Specimens of these plants were deposited in the National Herbarium of Addis Ababa University (AAU) for future reference.

The plant materials were cleaned thoroughly with tap water to remove contaminants and dried in shed at room temperature from 2 to 3 weeks in the Biomedical Sciences laboratory, AAU. The dried plant parts and resins were finely powdered using a kitchen grinder (Mortar and Pestle, sized about 9 inches in diameter). The powders were put through a sieve of 3 mesh sizes so as to filter a gross solid matter. The gel of *A. pulcherrima* plant was collected by spilling-out its fresh leaves with the knife. This gel is highly moisturizing content of the Aloe plant, which is used for various medicinal uses.

The extracts were prepared using a procedure similar to that often used by traditional healer's or patients, with some minor modifications. The plant powders were soaked in distilled water, which was placed on a shaker for 72 hr. The mixture was filtered through cotton/gauze, then through Whatman filter paper number 1 (pore size: 11μ m) to remove fine solid plant particles or insoluble constituents. The entire extracts were concentrated to dryness using lyophilizer under reduced pressure. Then, the semi-solid concentrates poured into a glass petri-plates and allowed to completely dry in water bath adjusted to 45°C. The final dried extracts were collected and stored in labeled sterile bottles covered with air-tight stopper and kept in freezer on -20°C until used in the experiments.

Phytochemical Screening

In the present study, a preliminary phytochemical screening of the selected medicinal plants of the aqueous crude extracts was carried out for qualitative estimations of phytoconstituents.²⁷ The presence of the following secondary metabolites was evaluated using the respective methods as alkaloids,^{28,29} phenol,³⁰ flavonoids,³¹ tannins,^{32,33} steroids,³⁴ terpenoids,³⁵ saponin,³⁶ Glycosides,³⁷ and steroidal glycosides.³⁴

Chemicals and Reagents

The chemicals used in the present work were analytical grade purchased from local distributors of Ethiopia. The procurement of the required supplies, chemicals and reagents including petroleum ether, chloroform, ethyl acetate, butanol and ethanol were from Wisteam PLC; DPPH (2,2-diphenyl-1-picrylhydrazyl) and sodium chloride (NaCl) were purchased from Micron International Trading House PLC. Ascorbic acid was obtained from Food Sciences Department of AAU, which was purchased from Micron International Trading House PLC, Addis Ababa, Ethiopia.

DPPH Radical Scavenging Activity

DPPH (2,2-diphenyl-1-picrylhydrazyl) free radical scavenging potentials of plant extracts were determined using similar methods to¹¹ and³⁸ with minor modifications. DPPH is commercially available radical capable of accepting an electron (hydrogen) to become stable molecule, with a maximum UV-Vis spectrophotometry absorption at 517nm for 30 min.^{18,39} DPPH colorimetric assay, the original deep violet color of DPPH solution converts into a stable yellow-white color in the presence of the potent antioxidants.²¹ The basic chemical reaction of DPPH in the presence of an electron donating antioxidant is represented as: DPPH + AH \Rightarrow DPPH-H + A^{*}.^{21,40}

The antioxidant activities of the various concentrations (0.20, 0.39, 0.78, 1.56, 3.13, 6.25, 12.50, 25, 50, 100mg/ml) of plant extracts compared to standard drugs were examined. Ascorbic acid was used as a reference drug. The working solution was prepared by diluting the stock solution to a series of concentrations. A solution of 0.5mM DPPH ($C_{18}H_{12}N_5O_{6}$, Molecular weight: 394.32) (0.19 mg/ml) in methanol was prepared. Then, 2ml of DPPH solution was mixed with 1ml of plant extract prepared in methanol. The mixture was vortexed thoroughly and left

in the dark chamber at room temperature for 30 min and measured spectrophotometrically at 517nm for light transmittance (absorbance). These were analyzed in a quartz cuvette (optical path: 10mm and size: 4ml) at room temperature. The ability of the plant extract to scavenge DPPH radical was calculated as follows.

DPPH radical scavenging activity (%) = [(Control - Sample) /Control] \times 100, where; Control refers to the absorbance of DPPH radical plus methanol; Sample = the absorbance of DPPH radical plus the extract/ Ascorbic acid. Methanol was used as a blank reference. Each test was performed in triplicate.

The antioxidant activities of *A. aspera*, *S. punctata*, *A. pulcherrima*, *G. fruticosus* and *C. myrrha* extracts were assessed at various concentrations ranging from 0.02 to 100mg/ml. According to the United States food and drug administration (FDA), the half maximal inhibitory concentration (IC₅₀) represents the concentration of a drug, which is required for 50% inhibition under *in vitro* test.^[41,52] The dose-response experiments were performed 3 times for each extract/standard control, and the mean IC₅₀ ± SD was calculated. The percent of control was calculated as absorbance units in the presence of Ascorbic acid.

Statistical Analysis

Data were analyzed using Graph Pad Prism version 6 Software (Graph Pad Software, San Diego, CA, USA). One-way analysis of variance (ANOVA) followed by Dunnett's comparison test was performed to compare between the positive control (ascorbic acid) and plant extracts groups for antioxidant evaluations. Nonlinear regression analysis was used to calculate the IC₅₀ values for all dose-response curves. The data values were expressed as mean \pm standard deviation (SD) of three replicates. Values of *p*<0.05 was considered statistically significant.

RESULTS

Phytochemical Analysis

Preliminary phytochemical analysis of the selected medicinal plant aqueous extracts resulted in the identification of the following constituents (Table 1).

DPPH radical scavenging assays

Ascorbic acid (vitamin C), the positive control revealed the highest DPPH radical scavenging activity with 87.6% reductions at concentration of 0.20 mg/ml, and 98.4% at 100 mg/ml concentration. The antioxidant activity of *S. punctata* extract was 92.3% inhibition at 0.20 mg/ml, which was higher than the effect of ascorbic acid (87.6%) at 0.20 mg/ml. The antioxidant activity of *S. punctata* at dose 100 mg/ml was 88.5%, which was lower than ascorbic acid (98.4%), although not statistically

significant. In general, *S. punctata* extract was a strong antioxidant equivalent to ascorbic acid. The antioxidant activity of *A. pulcherrima* extract was 72.3% at 0.78 mg/ml, which was significantly different from that of ascorbic acid (94.5%) (P<0.01). However, as the concentrations of *A. pulcherrima* extract increased beyond 6.25 mg/ml (88.4%), its antioxidant activity declined to 69.7% at 100 mg/ml significantly lower than ascorbic acid (P<0.001). *G. fruticosus* showed a scavenging potential of 81.6% at 3.13 mg/ml, which was significantly different from ascorbic acid (92.3%) (P<0.05). In addition, *G. fruticosus* scavenged free radicals by 93% at 12.50 mg/ml, which was not significantly different from the positive control, ascorbic acid (92.8%).

The antioxidant activity of *A. aspera* leaves (48.9%) at 12.50 mg/ml was significantly lower than ascorbic acid (95.6%) at the same concentration. Similarly, the antioxidant activity of *A. aspera* inflorescences was 54.9% at 6.25 mg/ml compared to ascorbic acid (94.1%) (*P*<0.001). DPPH scavenging potentials of *A. aspera* inflorescences was 71.6% compared to ascorbic acid (95.5%) at 25 mg/ml (*P*<0.001). The antioxidant activity of *A. aspera* leaves increased in a concentration dependent manner and it was 94.6% at 100 mg/ml equivalent to ascorbic acid (98.4%). However, *A. aspera* roots extract didn't possess antioxidant activity. DPPH radical scavenging activity of *C. myrrha* extract was 15.9% at 100 mg/ml, which was significantly lower than that of ascorbic acid (*P*<0.001). The IC50 antioxidant values of *S. punctata*, *A. pulcherrima*, *G. fruticosus*, *A. aspera* and *C. myrrha* extracts were 0.01 ±0.003 mg/ml, 0.42 ±0.047 mg/ml, 1.64 ±0.147 mg/ml, 13.51 ±1.08 mg/ml and nill, respectively, compared to Ascorbic acid (0.03 ±0.007 mg/ml) (Figure 1).

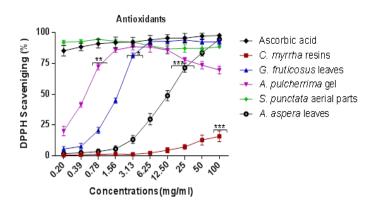


Figure 1: DPPH scavenging activities (%) of aqueous extracts of test plants at various concentrations. *C. myrrha, G. fruticosus, S. punctata, A. pulcherrima* and *A. aspera* extract as compared to the positive controls-ascorbic acid. Data were based on mean \pm SD of three independent observations. * *P* < 0.05, ** *P* < 0.01, *** *P* < 0.001 statistically significant compared to the control group.

Table 1: Phytochemical screening of the cr	ude aqueous extracts o	f medicinal plants.
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		Chemical constituents								
Extracts	Phenols	Flavonoids	Tannins	Steroids	Terpenoids	Steroidal Glycosides	Alkaloides	Saponins	Resins	Glycosides
A. aspera	+	+	+	+	+	+	+	+	_	_
S. punctata	+	+	+	+	+	+	-	+	+	-
A. pulcherrima	+	+	+	_	+	_	_	_	_	+
G. fruticosus	+	+	+	+	_	+	+	+	_	+
C. myrrha	_	_	_	+	+	+	_	_	_	+

DISCUSSION

The antioxidant constituents of plant extracts might restore the antioxidant enzymes of kidneys, which help to prevent renal cell injury.¹¹ Antioxidants have been reported to reconcile free radicals by directly reacting and quenching their catalytic metal ions.⁴² In other words, antioxidants neutralize the damaging effects of free radicals on tissues by donating electrons at the site of injury.⁴³ Previous studies reported that DPPH scavenging potentials of whole leaves extract of *Aloe ferox* was concentration dependent in which hydrogen donating compounds in the extract might increase.⁴⁴ In the present study, *S. punctata* extract has scavenged DPPH free radicals as postulated that it may react with hydrogen donors.¹³

Previous phytochemical studies on other plants had reported that flavonoids,45 saponins and terpenoids46 and phenols47 exhibited antioxidant activities. In line with these studies, S. punctata aqueous extract possesses phytochemicals including phenols, flavonoids, saponins and terpenoids showing antioxidant effects similar to ascorbic acid. Similarly, the aqueous extract of A. pulcherrima consists of phenols, flavonoids and terpenoids, whereas G. fruticosus extract possesses phenols and flavonoids, which shows DPPH radical scavenging activities. Therefore, targeting reductions in oxidative stress would be therapeutic options in the treatment of urolithiasis. The present in vitro study showed that the DPPH radical scavenging activity of half minimum inhibitory concentration (IC₅₀) of S. punctata, A. pulcherrima, G. fruticosus, A. aspera and C. myrrha aqueous extracts were 0.01 ±0.003 mg/ml, 0.42 ±0.047 mg/ml, 1.64 ±0.147 mg/ml, 13.51 ±1.08 mg/ml and nill, respectively. This indicates that S. punctata extract effectively scavenges DPPH radicals at the lower concentration than the other plants.

The antioxidant properties of test extracts may be potential contributors for lithotriptic effects,⁴⁸ and *S. punctata* aqueous extract was found to be the most potent antioxidant comparable to ascorbic acid in the present study. As reported earlier, the presence of antioxidant metabolites like flavonoids and phenols might be responsible for antiurolithiatic effects.⁴⁹ These were present in *S. punctata*, *A. pulcherrima* and *G. fruticosus* aqueous extracts phytochemical analysis that would further strengthen their use to treat urolithiasis-induced oxidative stress. Furthermore, the antiurolithiatic activities of *S. punctata*, *A. pulcherrima* and *G. fruticosus* aqueous extracts might be due to the disintegrations of mucoproteins by saponins and tannins found in the present study. The IR analysis of *G. fruticosus* fraction II revealed the presence of hydroxyl groups attached to the aromatic ring structures, which may be responsible for the antioxidant properties quenching free radicals as supported by earlier studies.^{50,51}

Furthermore, it was also reported that the steroidal constituents of medicinal plants possess antiurolithiatic activities, suggesting that macromolecules may inhibit CaOx nucleation in the human urine,⁵² which was found in both *S. punctata* and *G. fruticosus* aqueous extracts. Although our *in vitro* antioxidant data could not be extrapolated into *in vivo* data, it gave an insight for the activities of extracts. The limitation of this study may be the use of DPPH assay was used to test antioxidant activities, although different substances may involve different mechanisms.

CONCLUSION

S. punctata extract exhibited significant antioxidant activities equivalent to ascorbic acid, a drug currently in use. In general, *S. punctata* extract was a strong antioxidant equivalent to ascorbic acid (control). Further investigations of the total phenolic and flavonoid content of extracts

will be required. Moreover, the *in vivo* study will be required in order to substantiate the *in vitro* findings.

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CONFLICT OF INTEREST

The authors declare that they have no competing interests.

ABBREVIATIONS

IR: Infrared; **IC**₅₀: half minimum inhibitory concentration; **mg/kg**: milligram per kilogram; **DPPH:** 2, 2-diphenyl-1-picrylhydrazyl; **SD**: Standard deviation.

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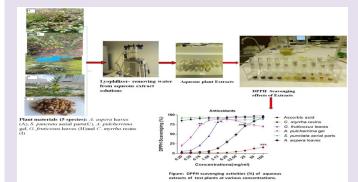
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SUMMARY

DPPH scavenging capacities of *S. punctata, G. fruticosus, A. pulcherrima* and *A. aspera* aqueous extracts were 92.3%, 81.6%, 72.3% and 54.9%, respectively compared to control (Ascorbic acid) showed 87.6%, 94.5%, 92.3% and 95.6%, respectively at inhibitory concentrations of 0.20 mg/ml, 0.78 mg/ml, 3.13 mg/ml and 12.5 mg/ml, respectively. The IC₅₀ antioxidant values of *S. punctata, A. pulcherrima, G. fruticosus,* and *A. aspera* extracts were 0.01 ±0.003 mg/ml, 0.42 ±0.047 mg/ml, 1.64 ±0.147 mg/ml, and 13.51 ±1.08 mg/ml, respectively compared to Ascorbic acid (0.03 ±0.007 mg/ml). Thus, the phytoconstituents in *S. punctata* aqueous extract has the best capability to scavenge DPPH free radicals.

GRAPHICAL ABSTRACT



ABOUT AUTHORS

Beyene Petros (Ph.D.) is a professor of Biology at Addis Ababa University. He attended elementary and high school at local schools in southern Ethiopia. He received his BSc from Addis Ababa University, MS from University of Wisconsin and Ph.D. from Tulane University all in Biology. Beyene joined the staff of Addis Ababa University in 1979 when he became a Lecturer. Later he became a Professor of Biology in 2009. He has more than 120 publications in peer reviewed journals. He has made successful contributions in diseases of public health importance and public health challenges.

Asfaw Debela (Ph.D.) is a researcher at the Ethiopia public health institute. He graduated from Addis Ababa University School of pharmacy with BA degree and obtained his M.Sc. in natural product chemistry. He has also received his Ph.D. in pharmaceutical chemistry from Karl- Franzens university of Graz, Austria, in 2001. He did his post doctoral work at institute of organic and medicinal chemistry, natural products and medicinal chemistry research unity, at Wuzbity University, Germany in 2008. He has published more than 86 scientific publications and co-authored one book. He has made successful contributions in problem solving researches for diseases of public health importance.

Tilahun Alelign has completed his B.Sc. at Bahir Dar University in 2005, and M.Sc. in Biomedical Sciences from Addis Ababa University, Ethiopia in 2010. He has 8 publications, and presented some of these papers at scientific conferences in Ethiopia. His research interests include searching a natural products to cure kidney stone, kidney failure, and related disorders. Currently, he is attending Ph.D. studies at Addis Ababa University, Ethiopia. His research project focuses on traditional medicine to come up with alternative treatment options for kidney calculi.

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